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contains the maximum amount of solute for a given quantity
of.... the amount of solute that dissolves in a given quantity of a
s.... a solution that contains less solute than a saturated
solution....

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Saturated Solution. A solution containing the maximum amount of solute for a given amount of solvent at a constant temperature and pressure; An equilibrium exists between undissolved solute and ions in a solution. Solubility.

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a. the reaction will shift left
b. the reaction will shift right
c. the reaction will not shift; it's at equilibrium
d. none of the above
5. If a reaction vessel contains a partial pressure of CO_2 at 0.50 atm and a partial

CHM 2046 Test #3 Review: Chapters 14.8, 14.9, 15, & 16

Chapter 16: Tests for ions and gases
Identification of cations (positive ions)
Identifying metal cations by using flame tests
The principle here is that the salts of some metals will impart a colour to a non-luminous Bunsen flame.

Chapter 16: Tests for ions and gases

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